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Key In chemistry,

an ionic compound

is a chemical

compound

composed of ions

held together by

electrostatic forces

termed ionic

bonding. The

compound is

neutral overall, but

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consists of
positively charged
ions called cations
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below. 3. Next to
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Ionic compounds

include salts,

oxides, hydroxides,

sulfides, and the

majority of

inorganic

compounds. Ionic

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Solids are held

together by the

electrostatic

attraction between

the positive and

negative ions. ions

electrons cations

accept electrons □

Students will

complete a cut and

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ionic bonds.

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Structure. The

structure of an

ionic compound

depends on the

relative sizes of the

cations and anions.

Ionic compounds

include salts,

oxides, hydroxides,

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solids are held together by the electrostatic

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Construct the
following ionic

compounds using
puzzle pieces that
you will cut out:

Copper (I)*

chloride. Copper

(II)* iodide.

Magnesium oxide.

Gold (III) Nitrate.

Sodium Sulfate.

Lead (II)*

Hydroxide.

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Aluminum
Carbonate. e. Iron
(III)* Phosphate.
Ammonium
Dichromate. Cut
out the appropriate
ion puzzle pieces.
Do not cut out all
the pieces on a
sheet.

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composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations

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and negatively charged ions called anions. These can

be simple ions such

as the sodium (Na^+) and chloride (Cl^-) in sodium

chloride, or

polyatomic species

such as the

ammonium ...

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