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solution, what will the concentration be if I add 560 mL more

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A 19 page worksheet and answers to all exercises are provided. This lesson is part of a series covering the OCR AS Chemistry specification and relates to the following sections: Module 2 – Foundations in chemistry. Part 1 – Atoms and reactions. 2.1.3 – Amount of substance.

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Given the following reaction: $(4) \text{H}_2\text{SO}_4 + \text{Na}_2\text{CO}_3 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O} + \text{CO}_2$. Calculate the molarity of the H_2SO_4 solution if it takes 40.0 mL of H_2SO_4 to neutralize 46.7 mL of a 0.364 M Na_2CO_3 solution.

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